



The Chemistry and Manufacture of Hydrogen

By Philip Litherland Teed

Theclassics.Us, United States, 2013. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can usually download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1919 edition. Excerpt: .metallic aluminium with mercury by hand, as advantage is taken of the fact that aluminium will reduce aqueous solutions of salts of mercury to the metallic state, in accordance with the following equation: $-2Al + 3HgCl_2 = aAlCl_3 + 3Hg$. Consequently, if there is an excess of aluminium over that required by the equation, this excess will be automatically amalgamated by the metallic mercury as it is produced. In a practical application of this method by Mauricheau Baupre, 1 fine aluminium filings are mixed with a small proportion of mercuric chloride ($HgCl_2$) and potassium cyanide (KCN), which causes a slight rise in temperature and produces a coarse powder, which is quite stable if kept free from moisture. This mixture can be kept in air- and water-tight boxes until it is required, when it can be gradually added to water kept at about 70 C. A brisk...



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